Magnetite -Zeolite Nanoparticles for extracting the *Dioxins (2,3,7,8- tetrachlorodibenzo para dioxin (TCDD)* in Polluted Waters: An In- Silico Study

Narendra Kulkarni¹, Dr Uday Muddapur¹, María Colín-García², Alejandro Heredia³

- KLE Society's Dr. M.S.Sheshgiri College of College of Engineering & Technology, Angol Main Road, Udyambag, Belgaum, Karnataka, India, Contact number :91-9164264699
- Instituto de Geologia, Universidad Nacional Autonoma de Mexico (UNAM), Ciudad Universitaria, 04510 Mexico, Mexico
- Departamento de Química de Radiaciones y Radioquímica, Instituto de Ciencias Nucleares, Universidad Nacional Autónoma de México, Circuito Exterior C.U. Apdo. Postal 70-543, 04510 México, D.F., México, Fax: (+52) 55-56-16-22-33

Contact e- mail id: narenkulkarni9@gmail.com

Dioxins are environmental pollutants. They are belonging to the "dirty dozen" - a group of dangerous chemicals known as persistent organic pollutants in water. Highly potential experiments have shown them affecting number of organs and systems. With different types of nanomaterial emerging for applications in water purification and water treatment devices show the effectiveness against both chemical and biological contaminants. Utilization of specific nanoparticles either embedded in membrane or on other structural media that can effectively, rapidly render unusable water to potable is being explored at a variety of sectors. Zeolite is a crystalline hydrated aluminosilicate of alkaline and earth metals. It is an effective sorbents and ion -exchange media for metal ions which was evaluated as an ion exchange media for the removal of heavy metals from acids mine waste water. Magnetite Nanoparticles embedded Zeolite is prepared for better treatment of Dioxins contaminated water which can make potable one for regular use. Magnetite particles can be synthesized by co-precipitation of iron (II) and iron (III) aqueous salts solution in alkaline medium .Because of strong magnetic dipole, magnetite particles tend to aggregate. Polymeric compounds with special functional groups or surfactants can be added into solution in order to stabilize the solution (Wormuth, 2001). This technology is having lots of advantages over methods but still needs lots of research before the successful industrial application.

References:

- 1. A.E. Alvarez, A.G. Sanchez and X. Querol, "Purification of metal electroplating waste waters using zeolites". Water Res., 37, 2003, 4855-4862.
- 2. B. Geng, Z. Jin, T. Li and X. Qi, "Kinetics of hexavalent chromium removal from water by chitosan-Fe0 nanoparticles". Chemosphere, 75 (6), 2009, 825-830.
- 3. Beck, A.J., Johnson, D.L. and Jones, K.C., The form and bioavailability of non-ionic organic chemicals in sewage sludge-amended agricultural soils. Science of the Total Environment, 185, 1996, 125-49.
- Fan, Z., Casey, F.X.M., Larsen, G.L. and HAKK, H., Fate and transport of 1278-TCDD, 1378-TCDD, and 1478-TCDD in soil-water systems. Science of the Total Environment, 371, 2006, 323-333.
- 5. "Dioxins and Dioxin-like Compounds in the Food Supply: Strategies to Decrease Exposure".
- 6. Mei Fang, Valter Ström, Richard T. Olsson, Lyubov Belova, K. V. Rao, Rapid mixing: A route to synthesize magnetite nanoparticles with high moment, Appl. Phys. Lett. 99, 222501, 2011.
- 7. Mei Fang, Valter Ström, Richard T. Olsson, Lyubov Belova, K. V. Rao, Particle size and magnetic properties dependence on growth temperature for rapid mixed co-precipitated magnetite nanoparticles, Nanotechnology, 2012, 23, 14, 145601.
- 8. Suslick K.S., Fang M., Hyeon T. Sonochemical Synthesis of Iron Colloids, J. Am. Chem. Soc., 118 (47): (1996) 11960–11961.
- 9. Woo K., Hong J., Choi S., Lee H.W., Ahn J.P., Kim C.S., Lee S.W. Easy Synthesis and Magnetic Properties of Iron Oxide Nanoparticles, Chem. Mater., (2004) 16:2814-2818.
- 10. Wormuth K.) Superparamagnetic Latex via Inverse Emulsion Polymerization, J Colloid and Interface Science, (2001, 241:366–377.